



Phytochemical Analyzes from the Leaves of *Bryophyllum pinnatum*

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Authors' contributions

This work was carried out in collaboration between all authors. Author EOF designed the study and wrote the first draft of the manuscript. Authors LW, SL, AGM and CAO managed the analyses of the study and edited the manuscript. All authors read and approved the final manuscript.

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ABSTRACT

Aims: To identify extractable components in the traditional medicinal plant, *Bryophyllum pinnatum* (BP).

Methodology: The leaves of BP were sequentially extracted with non-polar (CH₂Cl₂) and polar (CH₃OH) solvents. The extracts were derivatized to their volatile fatty acid methyl esters (FAME) and trimethylsilyl (TMS) esters/ethers and characterized by gas chromatography mass spectrometry (GC-MS).

Results: GC-MS analysis revealed the presence of sterols, fatty acids, monoarylphenolics, sugars, alcohols, sugar alcohols, sugar acids, carboxylic acids, dicarboxylic acids, tricarboxylic acids, vitamin, alkanes and alkenes.

Conclusion: Compounds identified (e.g. tocopherol) in the extracts are most likely responsible for its antimicrobial, antifungal, anticancer, antitumour and insecticidal activities.

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1. INTRODUCTION

Historically in Africa, people with diseases, ailments and other health conditions were treated by traditional healers who were versed in the use of herbs, (leaves, roots and barks), animals and marine products. In recent times, there has been an increase in the patronage of traditional medicine practitioners in Africa because of the dearth of orthodox medical doctors. For example in Nigeria, the doctor/patient ratio is 1:5000 [1] and this scenario is replicated in other African countries [2]. Another reason why many people are taking to natural medicine is that they are scared of the side effects in orthodox medicine, since it is generally believed that herbs are organic in nature and have no side effects, hence the preference for herbal treatment. Furthermore, the cost of accessing the services of traditional medical practitioners is inexpensive when compared to qualified medical doctors, which results in an increased patronage. Therefore, the use of medicinal plants, such as *Bryophyllum pinnatum* (*BP*), for treating illnesses is common in Africa.

BP, also known as *Kalanchoe pinnata*, is a perennial herb that grows up to 0.9-1.5 m tall, succulent dark green leaves and bell like pendulous flowers [3,4]. Common names for *BP* include: Zakhama-e-hyat, life plant, air or maternity plant, love plant, Canterbury bells, Cathedral bells, and parnabija. Moreover, *BP* is widely used in traditional medicine to treat ailments such as infections, healing wounds, and cancer. Its wide range of uses in folk medicine justifies its being called "life plant" or "resurrection plant", promoting researchers interest [5]. It grows in the wild and is used as folk medicine in tropical Africa, India, China, Australia, tropical America, Madagascar, Asia and Hawaii [6,7].

BP is astringent, sour in taste, sweet in the post digestive effect and has hot potency [8]. Furthermore, it is well known for its homeostatic and wound healing properties. The plant has gained considerable attention for its medicinal properties and finds application in folk medicine, as well as in the contemporary medicine [8,9]. In Southeastern Nigeria, the herb is used to facilitate the dropping of the placenta of a newly born baby [10,11]. The lightly roasted leaves are used externally for skin fungus and inflammations. The leaf infusion is an internal remedy for fever [12]. Its use in folk medicines is

for treatment of hypertension and kidney stones [6], pulmonary infections, and rheumatoid arthritis [13]. The plant has hepatoprotective activity and is also used to increase vascular integrity [7]. Leaf juice is used in the treatment of coughs, bronchial affections, blood dysentery, jaundice and gout [14]. In Madagascar, it is used in anthroposophical medicine to treat psychiatric disorders and as a tocolytic agent to prevent premature labor [15]. The leaf extract of *BP* is also suspected have anticancer, antihypertensive, antidiabetic, anthelmintic, antileishmanial and antioxidant activities [16].

Several classes of compounds have been identified in the extracts of *BP*. Sterols have been identified in *BP* namely; stigmasterol [17] and bryophyllol, bryophollone and bryophollenone [18] and they have been associated with anti-inflammatory and analgesic properties. Furthermore, the leaves contain bufadienolides, a class of steroids, which possess antibacterial and antitumor actions [8]. Work by Tatsimo et al. [5] found several kaempferol rhamnosides in the methanol extracts of *BP* that have antimicrobial activity. Waxes, alkanes and fatty alcohols have also been found [19,20]. Fatty acids (e.g. palmitic, stearic, arachidic and behenic acids) from *BP* were shown to have an immunosuppressive activity [21]. Di- and tri-acids (e.g. malic, succinic and isocitric acids) have also been observed in extracts and their pharmacological actions are unknown [8,22].

The aim of this study is to identify the yield and composition of the extracts of the leaves of *BP* to identify potential pharmaceutical candidates. Both a non-polar (dichloromethane, CH_2Cl_2) and polar (methanol, CH_3OH) solvent extracts were characterized by GC-MS as their trimethylsilyl (TMS) and fatty acid methyl esters (FAME) derivatives. FTIR spectroscopy was used to support the chemical functionality of the crude extract compounds.

2. MATERIALS AND METHODS

2.1 Collection of Plant Material

Bryophyllum pinnatum (*Lam.*) *Oken* leaves were collected from an open land on the campus of Obafemi Awolowo University (latitude: 7°27' 59.99" N and longitude: 4°33' 59.99" E), Ile-Ife, Nigeria. The leaves were identified and authenticated by Mr. Gabriel Ibanesebhor and a

voucher specimen deposited (Herbarium Ife number 17090) at the Department of Botany, Obafemi Awolowo University.

2.2 Sample Preparation and Extraction

The leaves were rinsed with water, air-dried and pulverized into a powder using a domestic blender and stored in an air-tight container. The moisture content (9.7%) of the sample was determined just before extraction. The sample (5 g, in duplicate) were Soxhlet extracted continuously overnight with CH₂Cl₂ (150 mL) for 24 h and then CH₃OH (130 mL) for 48 h. Both the CH₂Cl₂ and CH₃OH extracts were rotary evaporated to dryness under vacuum to yield 3.30% and 7.88% on a dry leave basis, respectively.

2.3 GC-MS of Extract TMS Derivatives

Extracts (1.0 mg, in duplicate) were weighed into GC vials to which CH₂Cl₂ (1 mL) containing anthracene as an internal standard (IS, 50 µg mL⁻¹) was added. The samples were silylated with addition of N,O-bis(trimethylsilyl)-trifluoroacetamide (BSTFA) containing 1% trimethylchlorosilane (TMCS) (50 µL) and pyridine (50 µL) and heated for 30 min at 70°C or continued heating until the solution became clear [23,24]. The prepared TMS derivatives were analyzed by GC-MS (FOCUS-ISQ, ThermoScientific) in the electron ionization mode; temperature profile: 40°C (1 min) ramped to 305°C (10 min) at 5°C min⁻¹; GC capillary column (RTx-5MS, 30 x 0.25 mm Φ, Restek). The eluted compounds were identified by their mass spectrum, spectral matching with the 2008 National Institute of Standard and Technology (NIST) spectral library and known standards.

2.4 GC-MS of Extract FAME Derivatives

Extracts (5.0 mg, in duplicates) were weighed into 5 mL reacti-vials™ to which CH₃OH/H₂SO₄/CHCl₃ (1.7:0.3:2.0 v/v/v, 2 mL) was added to each vial and heated for 90 min at 90°C. CHCl₃ contained 1-naphthaleneacetic acid as an internal standard (50 µg/mL) [23,24]. The mixture was allowed to cool after which water (1 mL) was added, shaken vigorously, centrifuged, the CHCl₃ layer removed, dried through anhydrous sodium sulfate and transferred to GC vials. The prepared FAME derivatives were analyzed by GC-MS (details given above) using a temperature profile of 40°C (1 min) to 320°C at

5°C min⁻¹. The eluted compounds were identified with authentic standards (C14 to C26 fatty acids) and by spectral matching with the NIST spectral library. The concentration of each compound was referenced against the area of internal standard.

2.5 FTIR Spectroscopic Analysis

The functional groups in the crude samples were determined by FTIR spectroscopy using a Nicolet iS5 spectrometer using a ZnSe attenuated total reflection (ATR) iD5 probe (ThermoScientific). Spectra were collected in duplicate; the absorbance spectra were baseline corrected and averaged using Omnic-v9 software (ThermoScientific) [24].

3. RESULTS AND DISCUSSION

3.1 GC-MS Analysis of the TMS Derivatized Extracts

To identify potential pharmaceutical candidates, the leaves of *BP* were extracted with CH₂Cl₂ and CH₃OH and the extracts subsequently derivatized to make the compounds volatile for analysis. The CH₂Cl₂ extract of *BP* was analyzed by GC-MS as their TMS derivatives as shown in Fig. 1a and Table 1. The analysis revealed the presence of 34 prominent peaks, which represent 32 different compounds distributed among ten 'groups' of organic compounds. The different compounds have the following concentrations alkanes (99 mg g⁻¹), sugar (25 mg g⁻¹), alcohols (31.4 mg g⁻¹), fatty acids (404 mg g⁻¹), sterols (165 mg g⁻¹), alkenes (16 mg g⁻¹), monoarylphenolics (16 mg g⁻¹), vitamins (9 mg g⁻¹) and unknown steroid (8 mg g⁻¹).

Glycerol which is a product of hydrolysis of triglycerides was detected as one of the major compounds present in the extract (25 mg g⁻¹). Fatty acids are common occurrence in plant leaves and have been identified in leaves of *Viscus album* [25]. Free fatty acids as a group have the greatest concentration in the extract with linolenic acid, palmitic acid and linoleic acid having 102, 111 and 112 mg g⁻¹ respectively. This is consistent with the findings in the literature [21,26,27]. The next group of compounds identified is sterols with a concentration of 165 mg g⁻¹. Among the sterols present in the extract is stigmasterol, this is in agreement with the study carried out by other researchers [17,18], isolated stigmasterol and other steroidal derivatives from *BP*. Another

group is alkanes with 99 mg g⁻¹ concentration in the extract and has been previously observed [20]. Vitamins and alkenes have the least concentration in the extract. Those in between are monoarylphenolic and alcohols with concentration of 16 mg/g and 31 mg g⁻¹, respectively. The peaks at the retention times of 10.36, 33.48 and 55.14 min were unknowns.

The *BP* leave CH₃OH extract was also analyzed by GC-MS as their TMS derivatives (Fig. 1b and Table 2). There were 31 prominent peaks identified in the analysis, which represent 27

different compounds distributed among ten 'groups' of organic compounds. The concentration of sugars in the extract is 503 mg g⁻¹ with disaccharides alone contributing 233 mg g⁻¹ of this concentration. The next class is sugar acids with the concentration of 189 mg/g of the extract of which 2-keto-gluconic acid has a concentration of 166 mg g⁻¹. Carboxylic acids are another group of compound that is contained in the extract with a total concentration of 144 mg g⁻¹. Monocarboxylic acid (4 mg g⁻¹), dicarboxylic acids (16 mg g⁻¹) and tricarboxylic acids (41 mg g⁻¹) were also detected and have been previously

Table 1. GC-MS analysis of *B. pinnatum* CH₂Cl₂ extract TMS derivatives

Peak no.	RT (min)	Class	Compounds	Molecular formula	M ⁺ (m/z)	Conc. (mg g ⁻¹)
1	10.28	ALC	Propan-1,2-diol TMS ₂ ether	C ₉ H ₂₄ O ₂ Si ₂	220	4.9
2	10.36		unknown		262	6.4
3	10.53	ALC	Butane-2,3-diol TMS ₂ ether	C ₁₀ H ₂₆ O ₂ Si ₂	234	6.7
4	17.29	SR	Glycerol TMS ₃	C ₁₂ H ₃₂ O ₃ Si ₃	308	25
	29.23		Anthracene (IS)	C ₁₄ H ₁₀	178	
5	30.09	ALK	Eicosyne	C ₂₀ H ₃₈	278	4.3
6	30.36	FA	Myristic acid TMS ester	C ₁₇ H ₃₆ O ₂ Si	300	7.6
7	32.51	ALC	Hexadecanol TMS ether	C ₁₉ H ₄₂ OSi	314	3.7
8	33.48		Unknown			14.0
9	34.12	FA	Palmitic acid, TMS ester	C ₁₉ H ₄₀ O ₂ Si	328	111
10	35.62	ALKE	9-Octadecenyl TMS ether	C ₂₁ H ₄₄ OSi	340	15.9
11	35.73	ST	(17a)-3-methoxyestra-1,3,5(10)-triene-14,17-diyl TMS ₂ ether	C ₂₅ H ₄₂ O ₃ Si ₂	446	4.3
12	36.46	ALC	Phytol TMS ether	C ₂₃ H ₄₈ OSi	368	5.3
13	37.05	FA	Linoleic acid TMS ester	C ₂₁ H ₄₀ O ₂ Si	352	112
14	37.17	FA	Oleic + linolenic acid, TMS ester	C ₂₁ H ₄₂ O ₂ Si	354 + 350	102
				C ₂₁ H ₃₈ O ₂ Si		
15	37.58	FA	Stearic acid, TMS ester	C ₂₁ H ₄₄ O ₂ Si	356	25.0
16	38.76	MP	3-Hydroxyanthranilic acid, TMS ₃	C ₁₆ H ₃₁ NO ₃ Si ₃	369	15.9
17	40.78	FA	Arachidic acid TMS ester	C ₂₃ H ₄₈ O ₂ Si	384	11.2
18	43.75	FA	Docosanoic acid TMS ester	C ₂₅ H ₅₂ O ₂ Si	412	13.3
19	46.51	FA	Tetracosonic acid TMS ester	C ₂₇ H ₅₆ O ₂ Si	440	12.2
20	47.27	ALK	Triacontane	C ₃₀ H ₆₂	422	10.3
21	47.48	ST	Estra-1,3,5-(10)-triene-16,17-diacetate,2-Nitro-3-TMS ether	C ₂₅ H ₃₅ NO ₇ Si	489	12.3
22	47.89	ALC	Hexacosanol, TMS ether	C ₂₉ H ₆₂ OSi	454	4.5
23	48.77	VIT	Tocopherol TMS ether	C ₃₁ H ₅₆ O ₂ Si	488	9.1
24	49.82	ALK	Dotriacontane	C ₃₂ H ₆₆	450	20.6
25	50.38	FA	Octacosanoic acid TMS ester	C ₃₁ H ₆₆ OSi	482	9.2
26	52.21	ALK	Tetracontane	C ₃₄ H ₇₀	478	56.8
27	52.63	ST	13,27-Cycloursan-3-one	C ₃₀ H ₄₈ O	424	11.4
28	52.71	ALC	Triaccontanol TMS ether	C ₃₃ H ₇₀ OSi	510	11.2
29	52.84	ST	Stigmasterol TMS ether	C ₃₂ H ₅₆ OSi	484	19.0
30	53.03	ST	Sitosterol TMS ether	C ₃₂ H ₅₈ OSi	486	46.3
31	54.45	ST	Lup-20 (29) -en-3-ol, acetate	C ₃₂ H ₅₂ O ₂	468	59.9
32	55.14	ST	unknown steroid			7.6
33	55.24	ALC	Dotriacontanol TMS ether	C ₃₅ H ₇₄ OSi	538	6.7
34	55.66	ST	Friedelan-3-one	C ₃₀ H ₅₀ O	426	12.0
Total						798

ALK: alkanes; SR: sugar; ALC: alcohols; FA: fatty acids; ST: sterols; ALKE: alkenes; MP: monoarylphenolics; VIT: vitamin; RT: retention time; MW: molecular weight; IS: internal standard.

Table 2. GC-MS analysis of *B. pinnatum* CH₃OH extract TMS derivatives

Peak no.	RT (min)	Class	Compound	Molecular formula	M ⁺ (m/z)	Conc. (mg g ⁻¹)
1	10.28	ALC	1,2 propanediolTMS ₂ ether	C ₉ H ₂₄ O ₂ Si ₂	220	5.6
2	10.53	ALC	2,3 butanediolTMS ₂ ether	C ₁₀ H ₂₆ O ₂ Si ₂	234	6.9
3	11.14	CA	Lactic acid TMS ₂	C ₉ H ₂₂ O ₃ Si ₂	234	3.8
4	15.29	DCA	Malonic acid TMS ₂ ester	C ₉ H ₂₀ O ₄ Si ₂	248	4.72
5	17.30	SR	Glycerol TMS ₃ ether	C ₁₂ H ₃₂ O ₃ Si ₃	308	68.4
6	18.20	DCA	Succinic acid TMS ₂	C ₁₀ H ₂₂ O ₄ Si ₂	262	11.1
7	18.82	SA	Glyceric acid TMS ₃	C ₁₂ H ₃₀ O ₄ Si ₃	322	6.27
8	22.81	DCA	Malic acidTMS ₃	C ₁₃ H ₃₀ O ₅ Si ₃	350	84.0
9	23.42	SALC	Threitol/erythritolTMS ₄	C ₁₆ H ₄₂ O ₄ Si ₄	410	9.09
10	24.27	SA	Threonic acid TMS ₄	C ₁₆ H ₄₀ O ₅ Si ₄	424	5.4
11	24.69	SA	ErythronicacidTMS ₄	C ₁₆ H ₄₀ O ₅ Si ₄	424	4.7
12	28.29	SALC	Arabitol TMS ₅ ether	C ₂₀ H ₅₂ O ₅ Si ₅	512	26.3
13	28.53	CA	unknown acidTMS ₂ ester	C ₁₅ H ₃₂ O ₄ Si ₂	322	7.0
14	28.63	TCA	Aconitic acid TMS ₃ ester	C ₁₅ H ₃₀ O ₆ Si ₃	390	6.8
	29.22		Anthracene (IS)	C ₁₄ H ₁₀	178	
15	30.06	SA	2-Keto-d-gluconic acid TMS ₅	C ₂₁ H ₅₀ O ₇ Si ₅	554	166
16	30.22	SR	Fructose TMS ₅ ether	C ₂₁ H ₅₂ O ₆ Si ₅	540	30.8
17	30.32	TCA	Isocitric acid TMS ₄ ester	C ₁₈ H ₄₀ O ₇ Si ₄	480	33.8
18	31.79	SR	GlucoseTMS ₅	C ₂₁ H ₅₂ O ₆ Si ₅	540	72.8
19	32.02	SR	Hexose TMS ₅	C ₂₁ H ₅₂ O ₆ Si ₅	540	13.3
20	32.55	SALC	HexitolTMS ₆	C ₂₄ H ₆₂ O ₆ Si ₆	614	23.9
21	32.85	MP	3,4,5 Trihydroxybenzoic acidTMS ₄	C ₁₉ H ₃₈ O ₅ Si ₄	458	25.4
22	33.62	SR	Glucose TMS ₅	C ₂₁ H ₅₂ O ₆ Si ₅	540	85.4
23	33.93	SA	Gluconic acidTMS ₆	C ₂₄ H ₆₀ O ₇ Si ₆	628	6.3
24	34.12	FA	Palmitic acid TMS ester	C ₁₉ H ₄₀ O ₂ Si	328	13.1
25	34.47		Unknown			32.7
26	35.46	Cyc	Myo-InositolTMS ₆	C ₂₄ H ₆₀ O ₆ Si ₆	612	16.8
27	37.32	SR	Disaccharide TMS ₈	C ₃₆ H ₈₆ O ₁₁ Si ₈	918	6.2
28	37.60	SR	Disaccharide TMS ₈	C ₃₆ H ₈₆ O ₁₁ Si ₈	918	10.2
29	42.99	SR	Disaccharide TMS ₈	C ₃₆ H ₈₆ O ₁₁ Si ₈	918	14.9
30	44.58	SR	Sucrose TMS ₈	C ₃₆ H ₈₆ O ₁₁ Si ₈	918	181
31	46.01	SR	Disaccharide TMS ₈	C ₃₆ H ₈₆ O ₁₁ Si ₈	918	20.4
Total						1003

ALC: alcohols; CA: carboxylic acids; DCA: dicarboxylic acids; TCA: tricarboxylic acids; SR: sugars; SA: sugar acids; SALC: sugar alcohols; MP: monoarylphenolics; FA: fatty acids; Cyc: cyclic compound;

observed [8,22]. Sugar alcohols are another constituent of the extract with 59 mg g⁻¹ and arabitol having the highest concentration (26 mg g⁻¹) in this group. Alcohols were a minor class of compounds detected (12 mg g⁻¹). Other groups of compounds in the extract include free fatty acids (13 mg g⁻¹), cyclitols (17 mg g⁻¹) and monoarylphenolics (25 mg g⁻¹). The peaks at retention times of 28.53 and 34.47 min were unidentified. The various types of sugars identified in the extract are consistent with the findings from flavonoid glycosides in *BP* [15], however, no flavonoids were detected.

A number of active compounds, including phenolics (3-hydroxyanthranillic acid), steroids (e.g. stigmasterol and sitosterol), and organic acids, have been identified in the *BP* extracts [28]. Other compounds (e.g. tocopherol) identified in the extracts are potentially responsible for its antimicrobial, antifungal,

anticancer, antitumour and insecticidal activities [28]. *BP* anti-inflammatory activity can also be attributed to the presence steroids detected in the extracts [17].

3.2 FAME Analysis of the Extracts

All the fatty acids present in both extracts were converted to volatile FAME derivatives for GC-MS analysis (Table 3). The CH₂Cl₂ extract revealed 17 peaks in which 13 of them were identified for various fatty acids ranging from C14 (myristic acid) to C26 (hexacosanoic acid) (Fig. 2a). Palmitic acid (210 mg g⁻¹), linoleic acid (268 mg g⁻¹), linolenic + oleic acid (266 mg g⁻¹) and stearic acid (57 mg g⁻¹) were the major types detected. The minor acids identified in the CH₂Cl₂ extract are myristic acid, eicosanoic acid, behenic acid, and lignoceric acid. Other fatty acids with lesser concentrations in the extract include pentadecanoic, heptadecanoic,

tricosanoic acid, and hexacosanoic acid. The only class of compound identified in the CH₂Cl₂ extract apart from fatty acids were three alcohols, 3,5,11,15-tetramethyl-2-hexadecen-1-ol (7 mg g⁻¹), phytol (57 mg g⁻¹) and icosen-1-ol (25 mg g⁻¹). Fatty alcohols have been previously observed in *BP* [8].

In the CH₃OH extract, 17 major peaks were revealed in the GC-MS analysis of the FAME derivatives (Fig. 2b, Table 3). There were 12 fatty acids identified in CH₃OH extract from myristic acid to lignoceric acid. Palmitic acid (123 mg g⁻¹), linoleic acid (72 mg g⁻¹), linolenic + oleic acids (75 mg g⁻¹) and stearic acid (23 mg g⁻¹) were the major acids in the extract. The minor FAMES and their concentrations were myristic acid, iecosanoic acid, behenic acid, tricosanoic acid, 2-hydroxy-docosanoic acid, lignoceric acid and 2-hydroxy-tetracosanoic acid. These fatty acids were also found in other naturally occurring plant product extracts [23,24,29]. Other non-fatty acids present in the extract were 3,5,11,15-tetramethyl-2-hexadecen-1-ol, phytol and icosen-1-ol with a total concentration of 21 mg g⁻¹. C14-C22 fatty acids were common to both extracts in varying concentrations. Pentadecanoic, heptadecanoic and hexacosanoic acids were present in the

CH₂Cl₂ but absent in the CH₃OH extract. On the other hand, 2-methoxyestra-1,3,5(10)-triene-3,16,17-triol, 2-hydroxy-docosanoic acid and 2-hydroxy-tetracosanoic acid were identified in CH₃OH extract and not present in the CH₂Cl₂ extract.

This study is consistent with the results obtained by other researchers [13,18,19] which have identified steroids, organic acids, phenolics and sugars from the leaves of *BP*. Other compounds identified in the extracts were alkanes and alkanols, and this is in agreement with the work of others [20,30]. Several fatty acids such as palmitic, linoleic, linolenic, stearic, lignoceric and behenic acids were detected in the extracts. Linoleic acid has the highest concentration in the CH₂Cl₂ extract while linolenic + oleic acids have the highest concentration in the CH₃OH extract. Almeida et al. [21] showed that fatty acids (palmitic, stearic, arachidic and behenic acids) from *BP* have an immunosuppressive effect *in-vivo*. Work by Agoramorthy et al. [31] has shown that fatty acids have antifungal and antibacterial properties. Studies by other researchers have shown that fatty acids are ubiquitous in plant species [23,24,29,31].

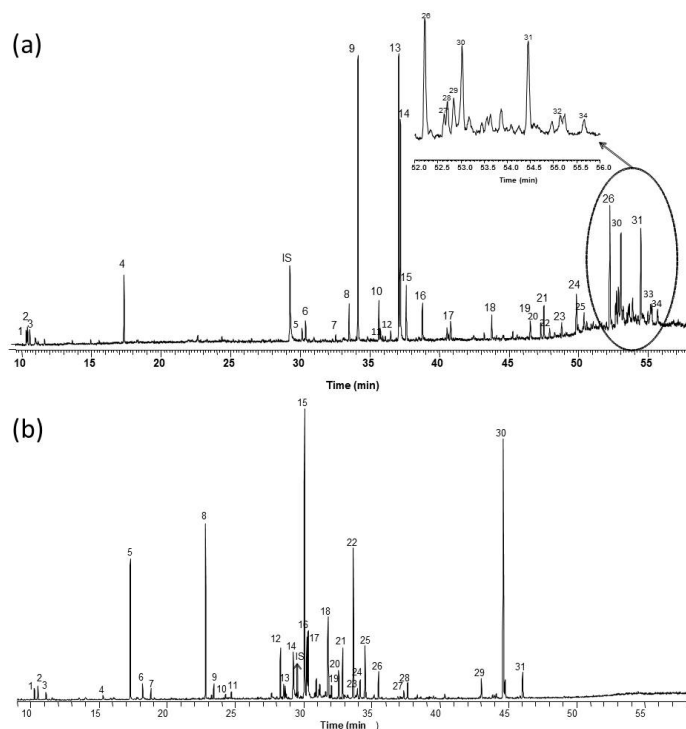


Fig. 1. Total ion current chromatogram of *B. pinnatum* TMS derivatives of (a) CH₂Cl₂ extract and (b) CH₃OH extract

Table 3. GC-MS analysis of FAME derivatives of *B. pinnatum* CH₂Cl₂ and CH₃OH extracts

Peak no.	Compound	RT (min)	M ⁺ (m/z)	CH ₂ Cl ₂ (mg g ⁻¹)	CH ₃ OH (mg g ⁻¹)
	Naphthaleneacetic acid (internal standard)	27.55	200		
1	Myristic acid (C14:0)	27.88	242	32.5	6.99
2	3,5,11,15 tetramethyl-2-hexadecen-1-ol	29.88	296	6.46	1.84
3	Pentadecanoic acid (C15:0)	29.99	256	2.0	
4	cisphytol	30.38	296	18.3	4.50
5	trans phytol	30.82	296	38.5	8.26
6	Icosen-1-ol	31.09	296	6.70	1.32
7	Icosen-1-ol	31.52	296	18.1	5.38
8	Palmitic acid (C16:0)	31.96	270	210	123
9	heptadecanoic acid (C17:0)	33.87	284	3.1	
10	Linoleic acid (C18:2)	35.11	294	268	72.3
11	Linolenic acid (C18:3) + oleic acid (C18:1)	35.24	292/296	266	75.2
12	Stearic acid (C18:0)	35.67	298	56.9	23.1
13	2-Methoxyestra-1,3,5(10) triene-3,16,17 triol	36.83	318		5.76
14	Eicosanoic acid (C20:0)	39.09	326	30.8	9.05
15	Behenic acid (C22:0)	42.23	354	39.0	14.9
16	Tricosanoic acid (C23:0)	43.77	368	2.6	1.93
17	2-Hydroxy-docosanoic acid (C22:0)	44.30	370		3.46
18	Lignoceric acid (C24:0)	45.58	382	32.6	6.09
19	2-Hydroxy-tetracosonic acid (C24:0)	48.72	398		7.20
20	Hexacosanoic acid (C26:0)	50.70	410	11.0	
Total				1043	370

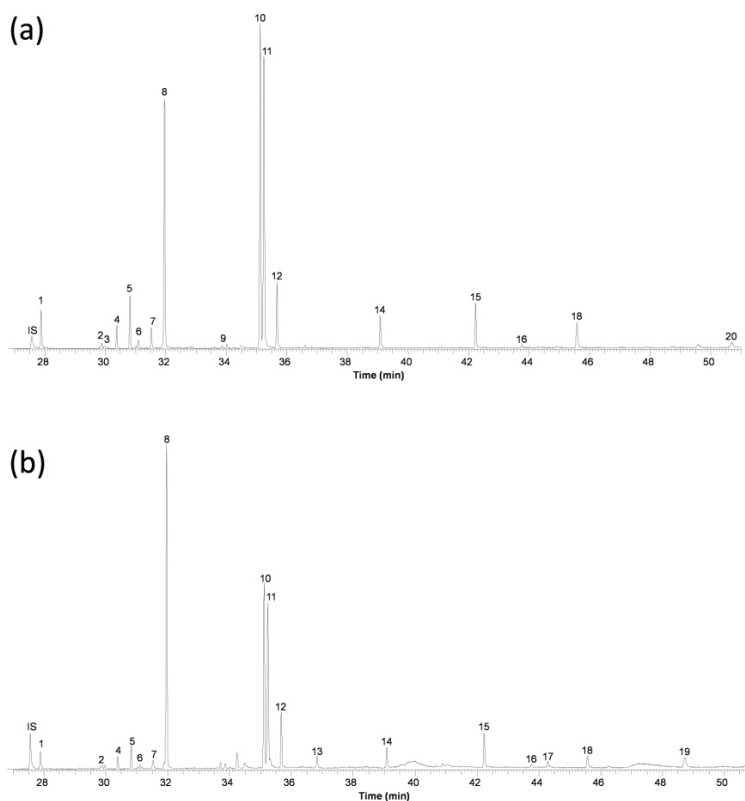


Fig. 2. Total ion current chromatogram of *B. Pinnatum* FAME derivatives of (a) CH₂Cl₂ and (b) CH₃OH extracts

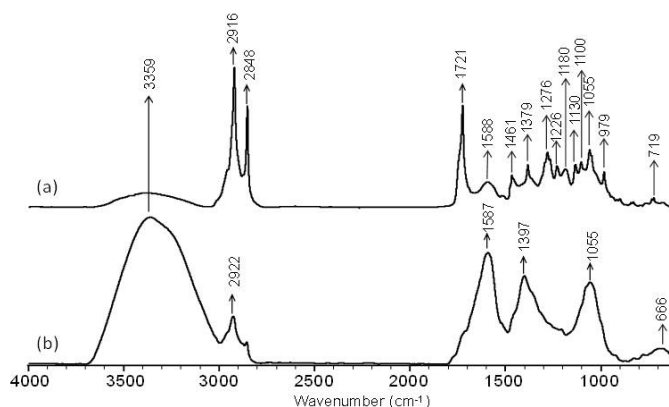


Fig. 3. FTIR spectra of *B. pinnatum* (A) CH_2Cl_2 and (B) CH_3OH extracts

3.3 FTIR Spectroscopic Analysis of *B. pinnatum* Extracts

FTIR spectroscopy was used to investigate the functional groups in the extracts. The spectra are shown in Fig. 3 above and the assignments of bands are discussed as follows. The presence of strong O-H stretching is observed at 3359 cm^{-1} corresponding to hydroxyl groups (Fig. 3B); however, carbonyl stretching of carboxylic acids was not seen at $\sim 1700\text{ cm}^{-1}$ [19,30]. Hence this O-H stretching band could be attributed to alcohols and sugars (see Table 2). The C-H stretching from methylene and methyl groups are shown at ~ 2922 and 2916 cm^{-1} , respectively, for both extracts and 2848 cm^{-1} only in the CH_2Cl_2 extract [32]. The pronounced carbonyl group band at 1721 cm^{-1} assigned to an ester was only observed in the CH_2Cl_2 extract. The intense bands at 1588 , 1587 cm^{-1} observed in both extracts were attributed to aromatic skeletal C=C vibrations of benzoic acid and other phenolics [32], however it is more pronounced in the CH_3OH extract than in the CH_2Cl_2 extract (Fig. 3A and Table 3). The observed band at 1461 cm^{-1} in the CH_2Cl_2 extract which was absent in the CH_3OH extract, was attributed to methylene scissoring. Methyl asymmetrical bending absorptions of methyl symmetrical bending were observed at 1379 cm^{-1} [32]. These spectral assignments are consistent with the FAME analysis results (Table 3). The intense band at 1068 cm^{-1} was assigned to C-O bonded stretching vibration of an alcohol [32,33].

4. CONCLUSION

Characterization of dichloromethane and methanol extracts from *B. pinnatum* has been

successfully carried out by means of GC-MS and FTIR analysis. The analysis revealed the presence of alkanes, alkene, carboxylic acids, dicarboxylic acids, tricarboxylic acids, sugars, sugar acids, alcohols, sugar alcohols, fatty acids, monoarylphenolics, steroids, vitamin and cyclic compounds. A number of active compounds, including phenols, steroids, and organic acids, have been identified in *BP* extracts. Compounds identified (e.g. tocopherol) in the extracts are most likely responsible for its antimicrobial, antifungal, anticancer, antitumour and insecticidal activities. *BP* anti-inflammatory activity can also be attributed to the presence of steroids. The present study shows the phytoconstituents of *BP*, which is very helpful to subsequent researchers to explore more into the potential benefits of this miracle plant.

CONSENT

It is not applicable.

ETHICAL APPROVAL

No animals or human test subjects were involved in this study.

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COMPETING INTERESTS

Authors have declared that no competing interests exist.

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