International Journal of Plant & Soil Science

31(3): 1-13, 2019; Article no.IJPSS.53537
ISSN: 2320-7035

Mineralization and Decomposition of Four Types of Compost Based on Biomass of *Sida cordifolia L.* in a Sandy Soil in the Semi-arid Zone of Niger

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Authors' contributions

This work was carried out in collaboration among all authors. Author SMS designed the study, performed the statistical analysis, wrote the protocol and wrote the first draft of the manuscript. Authors AKS and MB managed the analyses of the study. Author JBA managed the literature searches. All authors read and approved the final manuscript.

Article Information

DOI: 10.9734/IJPSS/2019/v31i330213
Editor(s):
(1) Mirza Hasanuzzaman, PhD, Associate Professor, Department of Agronomy, Faculty of Agriculture, Sher-e-Bangla Agricultural University, Dhaka-1207, Bangladesh.
Reviewers:
(1) Cristiane Ramos Vieira, Cuiabá University, USA.
(2) Moïse Yoni, Université de Dedeougou, Institut des Sciences de l'Environnement et du Développement Rural, Burkina Faso.
(3) Kouame Konan, University of Korhogo, Côte d'Ivoire.
Complete Peer review History: http://www.sdiarticle4.com/review-history/53537

Original Research Article

Received 25 October 2019
Accepted 27 December 2019
Published 03 January 2020

ABSTRACT

The low nutrient availability rainfall patterns regimes are the main constraints to agricultural production in Niger. This was a study of the decomposition and mineralization of nutrients of four types of composts (M1P, M2P, M1H and M2H) in a sandy soil. It was carried out at the experimental N'Dounga station (CERRA Kollo) located about 15 km from Niamey. A randomized blocks design with five repetitions was used. For the evaluation of yield, two doses (1 t ha^{-1} and 1.5 t ha^{-1}) were applied per millet. Decomposition and mineralization were assessed after burial at 10 cm depth between of a small bag containing 100 g (five small bags / compost). The characterization of the physico-chemical elements of composts samples after incubation has shown that composts are rich in nutrients. Nitrogen ranged from 0.8% to 1.1%, phosphorus from 9.99

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mg.kg^{-1} to 12.76 mg.kg^{-1} and potassium from 19.94 cmol$_c$ dm^{-3} to 26.26 cmol$_c$ dm^{-3}. All four composts are basic (pH> 7). Compost M2H lost more than 80% of its weight during the 10 weeks of the experiment compared to 48% for the M1P. the mineralization of N, P and K is greater at compost M1P (83.6% N, 72.72% P and 89.5% K). This compost also gave the highest yield (1272.5 kg ha^{-1}). The decomposition and mineralization of the main elements (N, P and K) allow the synchronization between the release of nutrients from these composts and the nutrient requirements of millet in a sandy soil.

Keywords: Compost; mineralization; Sida cordifolia; millet; sandy soil.

1. INTRODUCTION

Agricultural production in Sub-Saharan Africa is low and decling resulting from continued decline in soil fertility due to poor soil management and other biophysical factors [1].

In Niger, low crop yields are often explained by poor soil nutrient supply, unpredictable rainfall and low fertilizer use. This situation is aggravated by a population growth of about 3.8%, leading to frequent food shortages and persistent poverty in smallholder farming communities.

To increase yields in the Sahelian region of Niger, the use of mineral fertilizers is becoming increasingly necessary [2] where millet is a staple food and economic cereal for small farmers. The fertilizer recommendation on millet in Niger is 200 kg ha^{-1} of NPK compound (15-15-15) [3]. However, most farmers cannot afford to buy that amount of fertilizer. The high price of inorganic fertilizers, and the risks associated with their use in dry areas are the key factors limiting fertilizer use in Niger [4].

The option of integrated use of mineral and organic fertilizers to improve crop yields and maintain soil fertility is well documented [5,6,7]. However, the main sources of organic amendments such as crop residues and animal manure, are not available in sufficient quantities.

It is therefore necessary to develop another alternative low-cost soil fertility management option. The use of *Sida cordifolia* L., an herbaceous plant of the Malvaceae family as an organic material for the production of compost, is one of the alternatives to explore. To do this, an experimental study of composting the biomass of *Sida cordifolia* (BSC) was conducted in 2018. Four types of composts were developed to be applied by hill on millet. To be used as an organic amendment, these composts must be rich in nutrients but also, the decomposition and mineralization of nutrients must synchronize with the need of the crop.

The objective of this work is to evaluate the decomposition and nutrient mineralization of four types of composts based on the biomass of *Sida cordifolia* and their effects on millet yield in sandy soil.

2. MATERIALS AND METHODS

2.1 Study Site

The trial was conducted during the rainy 2017 season at the experimental station of the Regional Agricultural Research Centre (CERRA) of Kollo located at N'Dounga 15 km southeast away from Niamey. The station is located at a latitude of 13° 29'088'North and a longitude 2° 07'535'East (Fig. 1). The climate of the study area is of the Sahel-Sahelian type with an average annual rainfall of unclear. Average temperatures are around 30ºC in the dry season (March, April) and fall to 10°C in the harmattan season (December to February).

2.2 Preparation of Compost

Four types of compost were developed in 2018: M1P =Pit composting with 75% biomass of *Sida cordifolia*, 20% of organic manure and 5% Ash, M2P =Pit composting with 95% biomass of *Sida cordifolia* and 5% organic manure, M1H =Heap with 75% biomass of *Sida cordifolia*, 20% organic manure and 5% Ash, M2H =Heap composing with 95% biomass of *Sida cordifolia* and 5% of organic manure. The biomass of *Sida cordifolia* was harvested between October and November 2017 at maturity stage. In terms of organic manure and ash, they were collected in the village of Molli near the station where the composts were made.

2.3 Humidity, Temperature and Rainfall

At the beginning of the experiment, the maximum mean temperature was 33ºC and the minimum value of 22.4% was obtained 65 days after the test was installed (Fig. 2A).

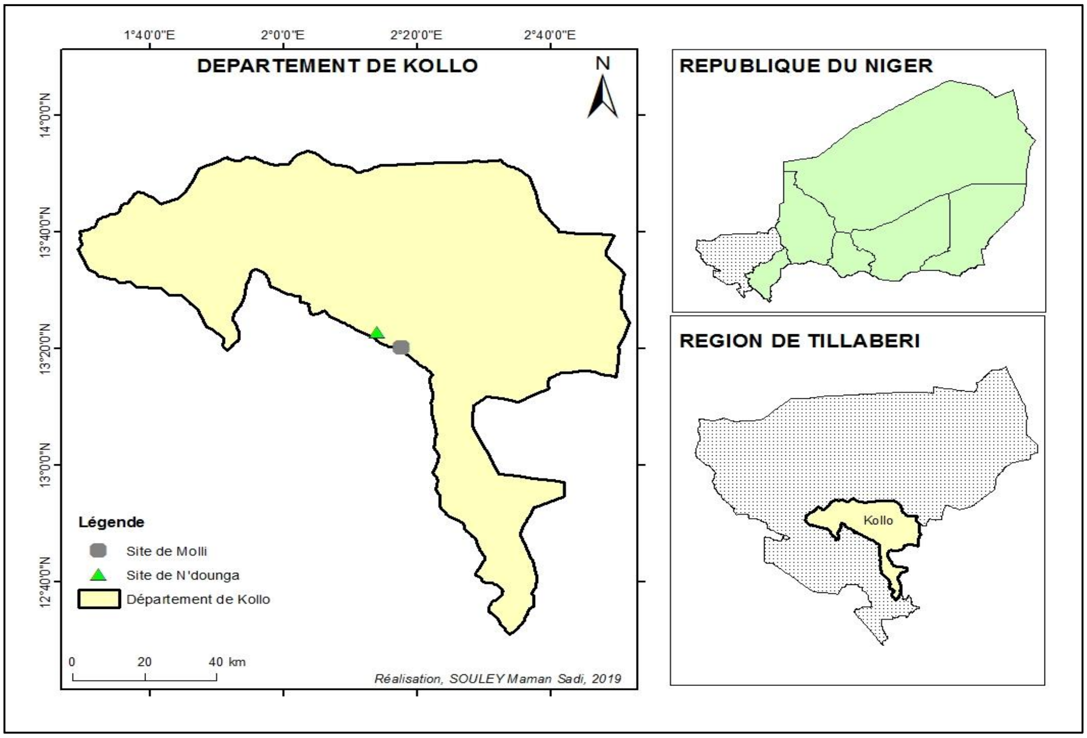

Fig. 1. Study site location map

Site moisture content ranged from 41.8% to 97.7% during the 10 weeks of the trial. The cumulative rainfall during the growing period was 579.4 mm. The highest rainfall (73 mm) was recorded on the 80th day after planting (Fig. 2B). Between the 30th and 59th day after the sowing (JAS), a small dry spell was observed. During this dry spell of 29 days the total rainfall was only 12 mm.

2.4 Physico-chemical Characterization of Composts and Soil

The physico-chemical characterization of the composts and soil was carried out by analyzing the physico-chemical elements on a composite sample of each compost. This analysis included the pH, C, N, P, and K.

The pH was measured according to the international standard ISO 10390 (1994) while the total organic carbon (TOC) was determined by the method [8]. The Kjeldahl method (NT 76.05, 1983) was used assessing nitrogen content. The method of [9] was used to determine available P content.

Potassium was determined using a flame photometer. The K^+ content was read directly into the mineralization.

Exchangeable bases (Na^+, Ca^{++}, Mg^+) were extracted by the ammonium acetate (NH_4OAc) solution at pH 7 using the extraction method described by Van Reeuwijk [10]. To determine the soil particle size, the Robinson pipette sampling method was used.

2.5 Experimental Design

The bag method was used to study decomposition of the compost. The litter bags measured of 20 cm x 15 cm and were produced from 1.0 mm nylon mesh.

The design was coupled with an experimental trial to evaluate the effects of *Sida cordifolia*-based composts on the yield of HKP millet. They were installed on 02 July 2018. Five (5) bags containing 100 g of each compost repeated 4 times were buried 10 cm deep. Before design installation, Physico-chemical characteristics of each compost were determined: dry weight, N, P, K and pH. Soil sampling was conducted to characterize the soil sites composition.

2.6 Data Collection

Random sampling of one bag of each compost and in each block was performed at 2, 4, 6, 8

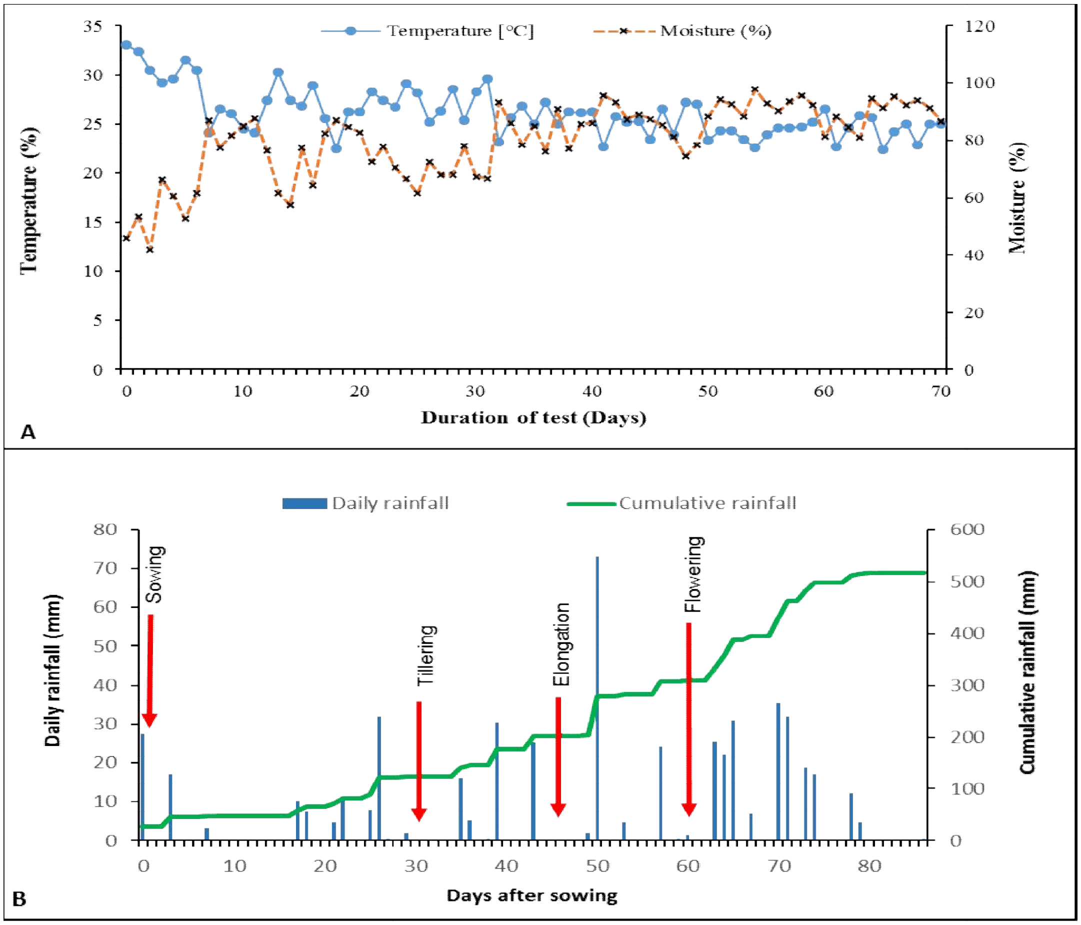

Fig. 2. Moisture and soil temperature (A) and rainfall distribution during the test period (B) at the site level

Source: http://www.fieldclimate.com INRAN REDSA ACC-3, Serial number 0020366

and 10 weeks. At each sampling, the remaining compost from each bag was manually cleaned and a fresh weight was recorded before drying it in an oven at 65ºC for 48 hours to take the dry weight.

The site's automated agro-metrological station was used to collect: moisture, temperature and rainfall data covering the trial period.

2.7 Statistical Analysis

A variance analysis at the 5% threshold was performed on data from the physico-chemical elements of composts. Each time a significant difference was detected, ANOVA was accompanied by the Fisher test (LSD). Excel 2016 and GENSTAT 9th edition software were used for all these analyses. The percentage of dry weight after sampling was determined using the formula:

$$\boldsymbol{Ps}\,(\%) = \mathbf{100\ x}\ \frac{\boldsymbol{P_t}}{\boldsymbol{P_0}} \qquad (1)$$

Where:

Ps (%) = Dry weight percentage;
Pt = Compost weight at t time;
P0 = Initial weight of compost in the bag.

The rate of release of nutrients following decomposition was calculated using the following formula:

$$\boldsymbol{T_n}\,(\%) = \mathbf{100\ x}\ \frac{C_0\,P_0 - C_t\,P_t}{C_0\,P_0} \qquad \mathbf{(2)}$$

Where:

Tn (%)=Nutrient Release Rate,
C0=Initial Concentration of Chemicals (N, P, K) from Compost,
Ct=Concentration of chemicals (N, P, K) compost at t time,

Pt=Weight of compost at t time, P0=Initial weight of compost in the bag.

The decomposition model and decomposition rate constant (k) of each type of compost were determined through the data that were modeled using a single exponential model described by [11]:

$$M_t = M_0 e^{-kt} \quad (3)$$

Where:

Mt = dry weight remaining of the compost at time t,

M0 = initial dry weight of the compost.

The time required for the compost to lose half its initial weight (t_{50}) was calculated using the formula described by Fening et al. [12]:

$$t_{50} = \frac{-\ln(0,5)}{k} \quad (4)$$

where k is the decomposition factor.

3. RESULTS

3.1 Physical and Chemical Composition of Soil at the Experimental Site

The results of the analysis of the soil samples (0-20 cm depth) collected at the site prior to the installation of the test (Table 2) showed that this is a sandy soil. The pH value was 6.77 available P content remained average (26.81 mg dm^{-3}). The N and C content of the soil is very low, 0.02% and 0.16% respectively. The low carbon and nitrogen content indicate that this soil has very soil fertility status (Table 1).

3.2 Physio-chemical Characteristics of Composts

The analysis showed clear differences with the composts with regard to OM (p<.001), N (p = 0.008), and K (p= <.001). No significant difference was observed for total and assimilated P and pH. The richest compost was the M1P compost containing 12.31% C, 1.11% N content and 26.6 $cmol_c$ dm^{-3}. The pit method gave a better quality than heap composting as this compost was richer in C, N and available P than the heap compost (Table 2).

3.3 Decomposition Model for Different Types of Composts

The decomposition pattern of different types of composts based on the *Sida cordifolia* biomass during the 2018 rainy season was shown in Fig. 2. The remaining weight of each decomposing compost was expressed as a percentage of the initial weight of the compost. The composts lost in average 59.4% its initial value after 10 weeks. Throughout the study, the M2H compost (heaps compost with 95% of SC+5% FM) decomposed faster than the others with a weight loss of 26% to 79% from the first month to the fourth month. There was little variation in the percentage of decomposition between the other composts (M1P, M2P, M1H). Within 30 days of the experiment, M1P, M2P and M1H composts lost 21%, 14% and 5% of their initial weights, respectively. At the end of the fourth month study the remaining amount of compost 48%, 52.6% and 58% for the M1P, M2P and M1H respectively (Fig. 3).

Table 1. Physical and chemical characteristics of the test site soil (n=4)

Measured parameters	Unit	Mean of values (depth 0-20 cm)
pH-H_2O (1 :2.5)		6.77±0.1
MO	%	0.28±0.02
C	%	0.16±0.01
Total N	%	0.02±0.002
Available P	mg dm^{-3}	12.81±8.11
Exchangeable bases		
CA^{++}	$cmol_c$ dm^{-3}	0.19±0.02
Mg^{++}	$cmol_c$ dm^{-3}	0.03±0.01
Na^{+}	$cmol_c$ dm^{-3}	0.31±0.04
K^{+}	$cmol_c$ dm^{-3}	1.02±0.47
Granulometry		
Clay	%	0.72±0.04
Silt	%	4.78±0.41
Sand	%	94.5±1.3

Legend: MO_- organic matter

Table 2. Mean composition in physio-chemical elements of composts (± SE)

Physico-chemical element	Types of composts				F.pr (0.05)	CV (%)	Norme	
	M1P	M2 P	M1H	M2H			FAO	AFNOR
pH-H2O (1 :2.5)	7.95 [a] (± 0.26)	7.58 [a] (± 0.85)	8.09 [a] (± 0.53)	8.15 [a] (± 0.51)	0.467	6.7		
Organic matter (%)	21.21 [b] (± 1.13)	19.16 [b] (± 0.88)	13.61 [a] (± 1.07)	14.21 [a] (± 1.33)	<.001	6.9	10-30	> 5
C (%)	12.31 [b] (± 0.65)	11.12 [b] (± 0.51)	7.90 [a] (± 0.62)	8.25 [a] (± 0.77)	<.001	6.9		
N (%)	1.11 [b] (± 0.14)	0.88 [a] (± 0.07)	0.80 [a] (± 0.08)	0.83 [a] (± 0.06)	0.008	11.3	0.4-0.5	> 0.25
tot P ($mg.kg^{-1}$)	122250 [a] (± 6850)	118000 [a] (± 4690)	125500 [a] (± 577)	126250 [a] (± 500)	0.086	3.5		
Available P ($mg.kg^{-1}$)	11.25 [a] (± 1.48)	12.76 [a] (± 1.46)	9.99 [a] (± 1.58)	10.07 [a] (± 1.55)	0.112	29.8		
K+ ($cmol_c\ dm^{-3}$)	26.26 [b] (± 0.32)	24.95 [b] (± 1.09)	19.94 [a] (± 1.31)	20.83 [a] (± 0.38)	<.001	4.2		
C/N	11.25 [a] (± 1.48)	12.76 [a] (± 1.46)	9.99 [a] (± 1.58)	10.07 [a] (± 1.55)	0.112	14.4	10-15	< 20

M1 P = Compost in pit with 75% SCB +20% OM+5% Ash, M2P = Compost in pit with 95% SCB +5% OM.
M1 H = Compost in heap with 75% SCB +20% OM+5% Ash, M2 H = Compost in heap with 95% SCB+5% OM.
FAO: World organization for agriculture and the food, AFNOR: Association French of Normalization.
tot P= total P and available P = Available phosphorus
Same letters within columns indicate no significant differences

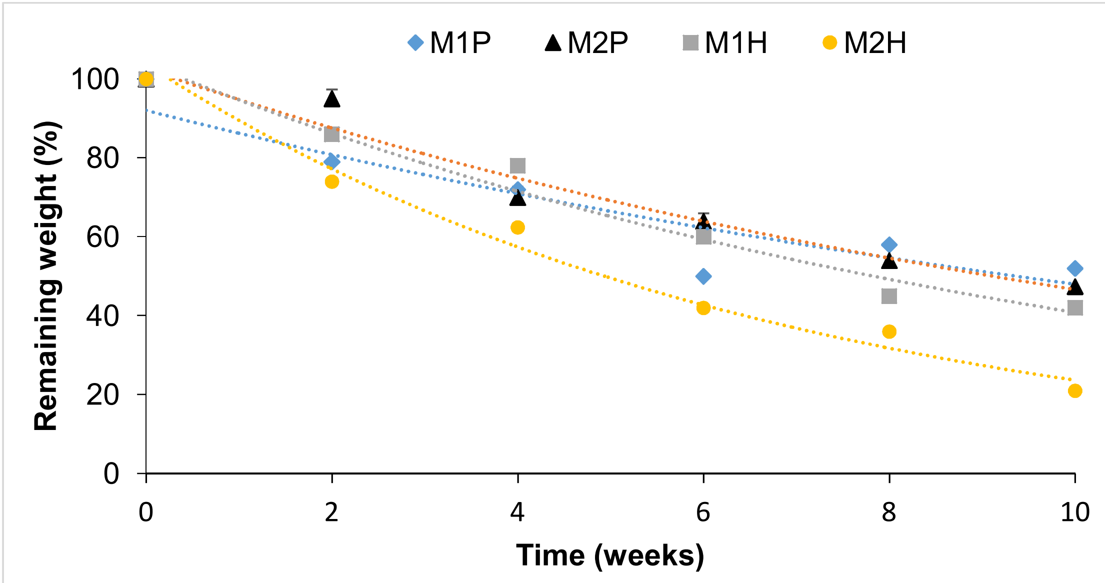

Fig. 3. Percentage of weight remaining in the litter bag

M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM. M1H=Compost in heap with 75% SCB +20% OM+5% Ash, M2H=Compost in heap with 95% SCB+5% OM Trend lines are the best fit

The composts differed greatly in number of weeks to lose half of its initial weight (t50). The two heap compost M2H and M1H had a t50 value of 2.33 and 3.85 weeks while the two peat compost M1P and M2P had t_{50} values of 5.33 and 4.33 respectively. The higher k value of the heap compost also reflected the higher decomposition rates for the heap composts as compared to the pit composts (Table 3).

3.4 Model for Nutrient Liberalization of Different Composts

3.4.1 Mineralization of nitrogen

None of the compost reached a 50% nitrogen liberalization during the first two weeks of the experiment, but the M1P compost which is richer in N release a larger part of its N than the other composts. This trend continued until the end of the experiment. At the end of the trial the compost at lost in average 76.5% of its initial value.

At the end of the study period, M1P compost released 83.6% of its N, while M2P, M1H and M2H composts released 66.6%, 74.8% and 81.02% of the N content, respectively (Fig. 4).

3.4.2 Mineralization of available phosphorus

During the first four weeks, the release of available phosphorus was slow. None of the compost released 50% of its available phosphorus.

Within 10 weeks, 72.72%; 69.17%; 68.47% and 67.68% of available P were released from M1P, M2P, M1H and M2H composts giving and average loss during the 10-week period of 69.5% (Fig. 5).

Table 3. Decomposition rate constant (k), coefficient of determination (R2) and (t50) of composts

Type of compost	Regression equation	k (day^{-1})	R²	T_{50}
M1P	M_t = 104,79$e^{-0,13t}$	0,13	0,81	5,33
M2P	M_t = 119,99$e^{-0,158t}$	0,16	0,97	4,33
M1H	M_t = 125,22$e^{-0,187t}$	0,18	0,97	3,85
M2H	M_t = 139,42$e^{-0,296t}$	0,29	0,97	2,39

Mt = compost remaining weight at t time. M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM.
M1 H=Compost in heap with 75% SCB +20% OM+5% Ash, M2 H=Compost in heap with 95% SCB+5% OM

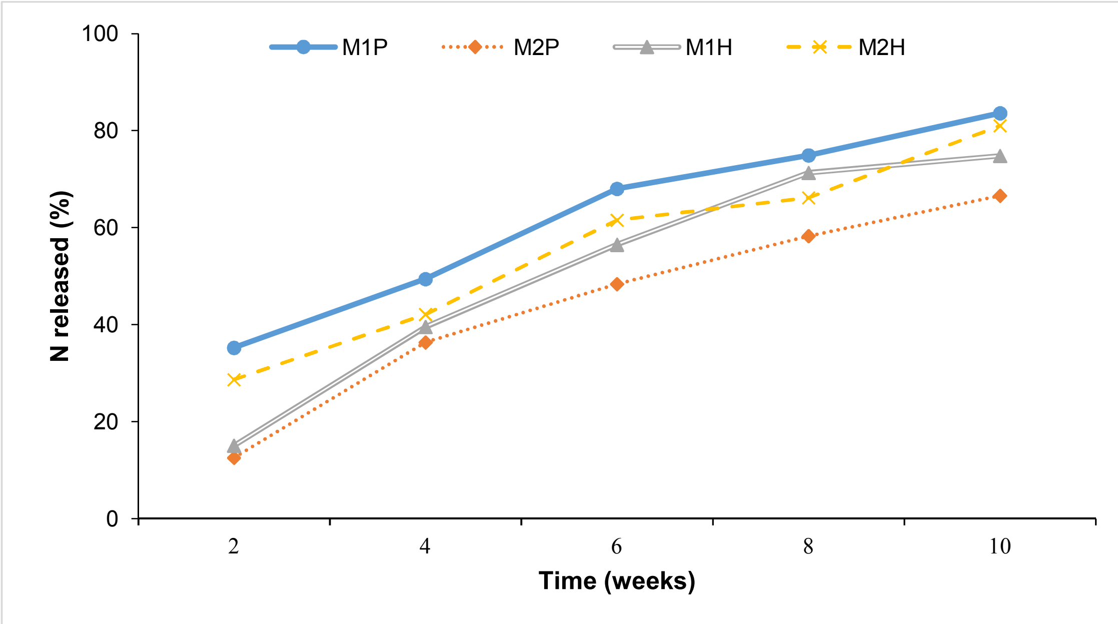

Fig. 4. Proportion of nitrogen released from different composts over time
M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM. M1 H=Compost in heap with 75% SCB +20% OM+5% Ash, M2 H=Compost in heap with 95% SCB+5% OM

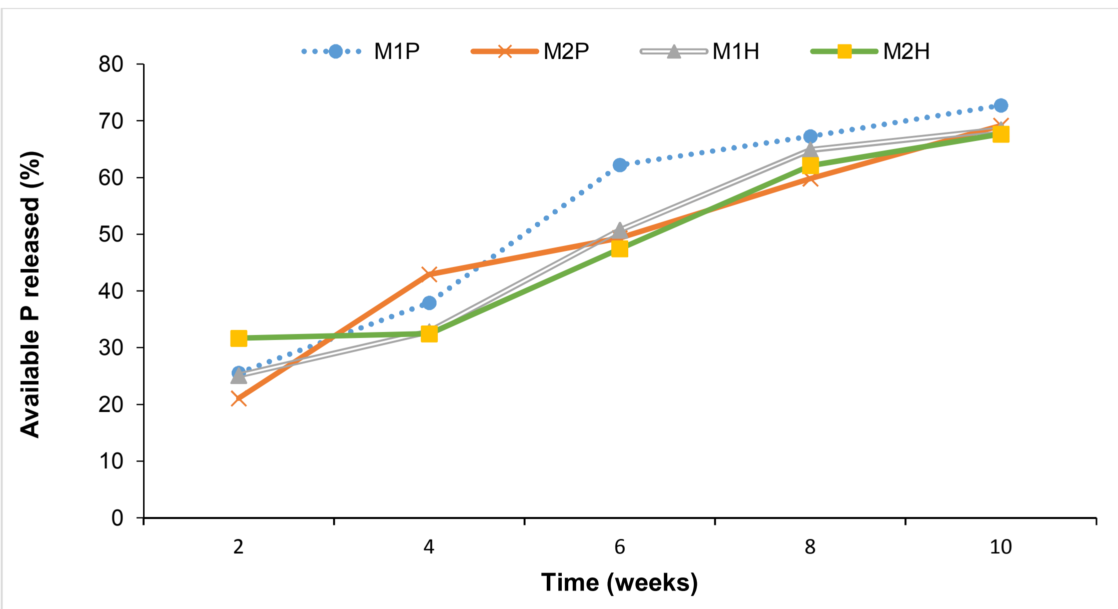

Fig. 5. Proportion of available phosphorus from different composts released over time
M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM. M1 H=Compost in heap with 75% SCB +20% OM+5% Ash, M2 H=Compost in heap with 95% SCB+5% OM

3.4.3 Mineralization of potassium

In this study, K mineralization was rapid as early as the second week for M1P and M2P compost, which released 56.98% and 52.02% of K respectively. The average loss in K during the 10-week period was 77.3% (Fig. 6).

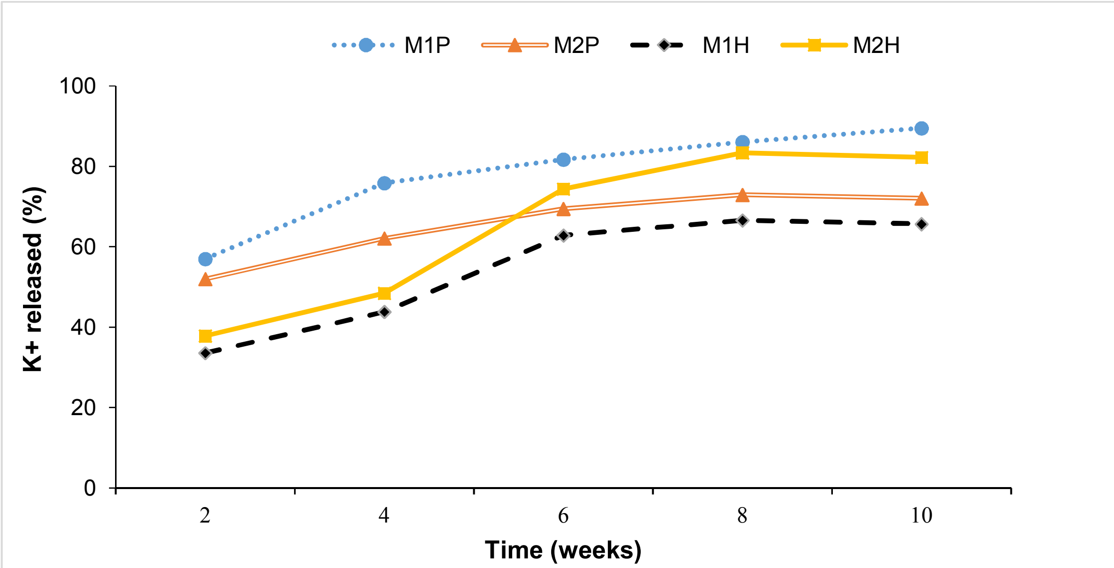

Fig. 6. Proportion of potassium released from different composts released 10 weeks

M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM. M1 H=Compost in heap with 75% SCB +20% OM+5% Ash, M2 H=Compost in heap with 95% SCB+5% OM

3.4.4 Mineralization of carbon

The release of C from the different composts ranged from 18% to 35% in the second week of the experiment. Only the composts M2H and M1H released more than 50% of the carbon during the first 6 weeks. After10 weeks, the compost M2H had the highest release of C and for this compost 84.2% of the carbon was released The compost M2P had the lowest loss of C and this compost lost 58.4% of its C after 10 weeks. The compost lost in average 69,0% of its initial value during the 10 weeks' period (Fig. 7).

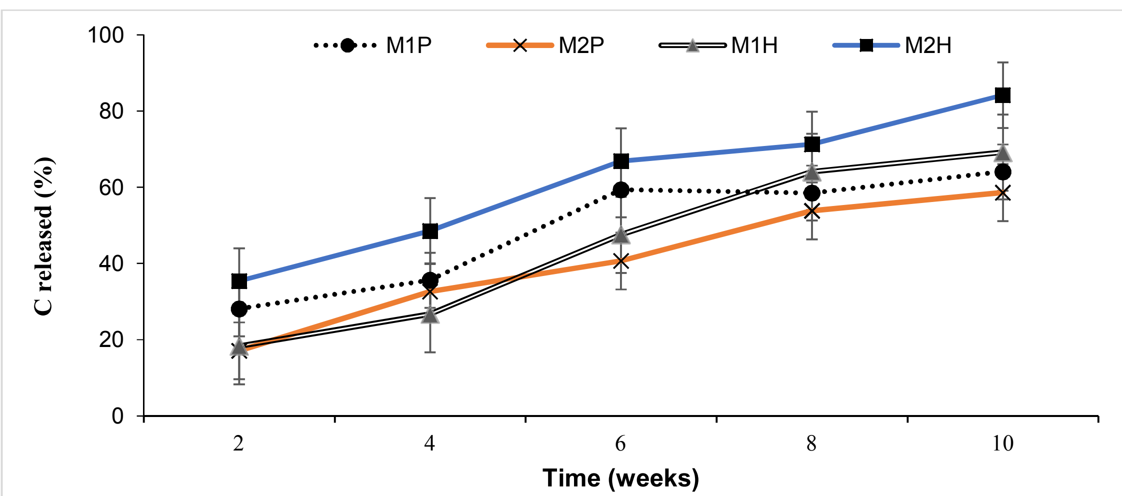

Fig. 7. Proportion of total carbon from different composts released over time
M1P =Compost in pit with 75% SCB +20% OM+5% Ash, M2P=Compost in pit with 95% SCB +5% OM. M1 H=Compost in heap with 75% SCB +20% OM+5% Ash, M2 H=Compost in heap with 95% SCB+5% OM

3.4.5 Effects of composts on millet grain yield and biomass

Analysis of variance showed effects of treatments on grain and biomass yields (P=<.001). The highest grain yields were obtained with the 1t. ha^{-1} and 1.5t. ha^{-1} doses of the M1P compost. This compost applied at the 1t. ha^{-1} dose increased the grain yield by 652 kg. ha^{-1} (105.2%) compared to the control.

For biomass yield, only M1P compost at 1t. ha^{-1} increased the yield compared to the control (1,377 kg ha^{-1}) (Table 4).

4. DISCUSSIONS

The study assessed the decomposition and nutrient mineralization of four types of compost based on the biomass of *Sida cordifolia* under field conditions and their effects on millet yield. In general, the composts are of good quality because their N and K contents were much richer than in the soil they are supposed to fertilize. Soil analysis showed that the soil N content was 0.016%, while the composts had a N content of between 0.8 and 1.1% (Table 2). The low soil N content, that was not been frozen before analysis, indicates that the soil has a very low capacity to supply nitrogen. The pit method gave a better quality than heap composting as this compost was richer in C, N and available P than the heap compost.

The decomposition study showed that between 50 and 80% of the plant nutrients contained in the composts were released within the 10 weeks' period. This shows that these composts can be used as fertilizer as this period corresponds to the period with high nutrient demand of the cereal crops. In this study pearl millet flowered after 8.5 weeks (60 days) (Fig. 2b). Cereals take up most of the nutrients during the period from sowing to flowering. There was a gradual release of plant nutrients during the 10 weeks (70 days) period even though there was considerable variation in soil water during the period of the experiment (Fig. 2). Even though these composts may not release their plant nutrients very early in a growing season, this may not be of great importance since it has been previously shown that fertilization in pearl millet in Niger can be delayed until 20 days after sowing without causing a yield penalty [3]. It is likely that nutrients reserve in the seed was sufficient in the first days after germination.

There was a variation in release pattern of the plant nutrients contained in the composts and the weight loss did not correspond to the nutrient release. The average loss with regard to weight, N, available P, K and C was in average 59.4, 76.5, 69.5, 77.3 and 69.0% respectively. The weight loss of the composts was therefore less than the nutrients loss. The mechanisms causing weight loss and nutrient loss are different explaining there was discrepancy between weight loss and nutrient loss. The losses were highest for nitrogen and potassium. This can be explained by the fact that these cations (NH^{4+} and K^+) are easily leached from the soil while the particles in the compost are likely to be more resistant degrading forces. Because, NH^{4+} and K^+ were not chemically strongly bound to the soil [13]. Therefore, these cations were exposed to leaching.

Table 4. Mean yield in grains and in stover of HKP millet according to treatments and sites

Treatments	**Yield (kg.ha^{-1})**	
	Grain	**Stover**
Control	620 a	1776 a
1t.ha^{-1} M_1P	1272.5 e	3153 b
1.5t.ha^{-1} M_1P	1127.5 e	2836 b
1t. ha^{-1} M_2P	856 cd	1500 a
1.5t. ha^{-1} M_2P	790 bcd	1565 a
1. ha^{-1} M_1H	832.8 bcd	1563 a
1.5t. ha^{-1} M_1H	871.8 cd	1635 a
1t. ha^{-1} M_2H	692 abc	1835 a
1.5t. ha^{-1} M_1H	685 ab	1517 a
Probability	**<.001**	**<.001**
CV	**12.1**	**25.3**

M1 P = Compost in pit with 75% SCB +20% OM+5% Ash, M2P = Compost in pit with 95% SCB +5% OM.
M1 H = Compost in heap with 75% SCB +20% OM+5% Ash, M2 H = Compost in heap with 95% SCB+5% OM.
Same letters within columns indicate no significant differences

The heap composts have higher weight losses than the composts produced in the pit as the heap composts lost half its weight in 3.12 weeks while the corresponding figure for pit compost was 4.83 weeks. The physical structure and resistance to degrading forcing may not be the same in pit and heap composting as there will be difference in temperature and water conditions between pit and heap composting.

The carbon loss during the 10 weeks' period was in average 69%. Even though the carbon decomposition will still continue for a few weeks, it is likely that not all the carbon supplied in the compost will be lost. The remaining carbon is of great value for building the soil organic carbon content therefore, there was an improvement of OM in the soil.

The composts produced can supply considerable amount of N. The average content in the compost was 0.90% corresponding to N input of 9 kg ha^{-1}. The litter bag study showed that 78.5% was released during the first 10 weeks. If this is taken into consideration the amount of easily available N will be 6.8 kg ha^{-1}. The recommendation by Tabo et al. [14] is to apply 2 g diammonim phosphate $hill^{-1}$ corresponding to an N input of 3.6 kg ha^{-1}. It is therefore clear that the N input from micro dosing of compost will be higher than the N input from micro dosing diammoium phospate.

With regard to P input the *Sida cordifolia* compost can also provide a substantial amount of P. The average available P content was 11.0 mg kg^{-1} compost (Table 2). Application of 1000 kg compost ha^{-1} (100 g per hill) will apply 7.6 kg ha^{-1} (1000 kg*0.011*69.5%) (69.5% is percent P released during the decomposition process) (Fig. 5). Application of 2 g diammonim $hill^{-1}$ as fertilizer will apply 4.0 ha^{-1} (20 kg fertilizer/ha*46% P_2O_5*0.436) which is almost half the amount of phosphorous applied with 100 g compost $hill^{-1}$.

The study also showed that a compost containing 95% *Sida cordifolia* and 5% manure can produce a compost of good quality as shown in the nutrient analysis of compost (Table 2). This is an indication that it will be easy to produce a compost were the main ingredient is *Sida cordifolia* since it is easily available in the agro-pastoral areas in Niger. This can also stimulate the farmers to use *Sida cordifolia* as a weed control. The C/N ratio of the four composts were close to 10 which is ideal value for a compost according to the norms of FAO. A C/N ratio of 10 implies that this is well decomposed compost than can be applied to the soil without causing N immobilization.

The application of composts improved millet yield over controls. The largest grain and biomass yield increase was achieved with M1P compost. The study of the mineralization showed that this compost released 83.6% of N, 72.72% of the available P and 89.5% of the K in 3 months. This could explain the increase in yield seen with this compost. Improved crop yield resulting from composting may be related to better crop development due to increased availability of nutrients from compost [15,16]. Studies by Esse et al. [17], Fatondji et al. [18] have also highlighted improved grain and biomass yields of millet under organic fertilization due to the progressive availability of nutrients for plants. In addition, M1P compost was the richest in nutrients compared to other composts. The dose of 1t ha^{-1} compost M1P was equivalent to the application of 1.11 g N per pouch corresponding to 11.1 kg of N ha^{-1} calculated on the basis of the density of the seedling of 10,000 feet / ha. This compost is also richer in OM (21.21%) and C (12.31). The use of this compost could durably improve the physical properties of the soil [19] reported that soil organic C is a sustainable land management index.

The study showed that the composts produced from *Sida cordifolia* are rich in plant nutrients and the plant nutrients were release gradually after incorporation in the soil. The weight loss of the composts were in average 59.4% during the 10-week test period while the corresponding release of N, P, and K was in 76.5, 69.5, and 77.3 respectively. This show that the release of these nutrients are well synchronized with nutrient demand in pearl millet as this crop reached flowering after 8.5 weeks. The plant nutrient release from the composts were highest for nitrogen and potassium as these plant nutrients are not strongly chemically bound in the soil.

5. CONCLUSION

The composts can be a good source of plant nutrients as 1000 kg compost ha^{-1} applied as micro dosing will apply more N and P than applied in 2 g diammonium phosphate $hill^{-1}$ corresponding to 20 kg DAP ha^{-1}. The *Sida cordifolia* mulch will over time improve soil OM a considerable amount of carbon remained at the end of the test period. Use of *Sida cordifolia* for compost production will not only increase soil fertility, but will also stimulate the farmers to cut this invasive weed species.

The contribution here, was to help farmers valorize the invasive weed to soil amendment and therefore increase their yield. This is an opportunity of preparing and selling compost to other farmers. This will allow to reduce use of chemical or inorganic and expensive fertilizer.

ACKNOWLEDGEMENTS

The research was funded by the Norwegian Ministry of Foreign Affairs through Norwewegian embassy in Bamako, Mali.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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